

Chapter 2, Lesson 2 Activity Sheet Answers

1. It is important to use the same amount of water on both paper towels because the experiment is trying to test whether *temperature* affects the rate of evaporation. The only thing different between the two samples of water should be the temperature they are exposed to. It wouldn't make sense to use different amounts of water because different amounts could affect how long it takes for the water to evaporate.
2. In an experiment, the samples being used should be treated the exact same way except for the one thing being tested. If you put one sample on the plastic bag with room temperature water and waited before putting the other on the plastic bag with hot water, this would not be a fair test. The samples would be on different surfaces when they should be treated the same way as much as possible.
3. Adding energy does increase the rate of evaporation. The evidence is that the water placed on the hot bag evaporated faster than the same amount of water placed on the room temperature bag.
4. Adding energy to the water on the paper towel makes the water molecules move faster. This motion competes with the attraction between water molecules and causes more of them to break away and evaporate.
5. Even though the wet paper towel was not heated water was able to evaporate because some water molecules were moving fast enough to evaporate.
6. When water evaporates, whole water molecules separate from other whole water molecules. The molecules do not come apart into their individual atoms.